

High Precision Osmium Isotope Analyses

Negative Thermal Ionization and Static Multiple Ion Collection

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Key Words

- TI-MS
- Osmium
- Negative Ions
- Faraday Mode
- Static Mode
- Finnigan™ TRITON

The rhenium-osmium geochronometer (beta-decay of ^{187}Re to the stable ^{187}Os) is of significant interest to the geochemical community. Early on, broad use of this chronometer was hindered by the lack of sensitive techniques to precisely measure isotopic abundances from small osmium samples.

In 1989, Heumann et al. (University of Regensburg, Germany) pioneered the possibility to measure osmium isotopic ratios with NTI-MS (Negative Thermal Ionization Mass Spectrometry). Shortly after this, the group of Wasserburg et al. (Caltech, USA) and others have improved chemistry and sample loading, with the result that this technique has evolved into an established tool for use by the geological community.

NTI is a powerful ionization technique for a broad range of elements. Many of the transition metals form negatively charged oxide ions. High ion yields can be obtained for elements with high electron affinities. The ion current is essentially influenced by the electron work

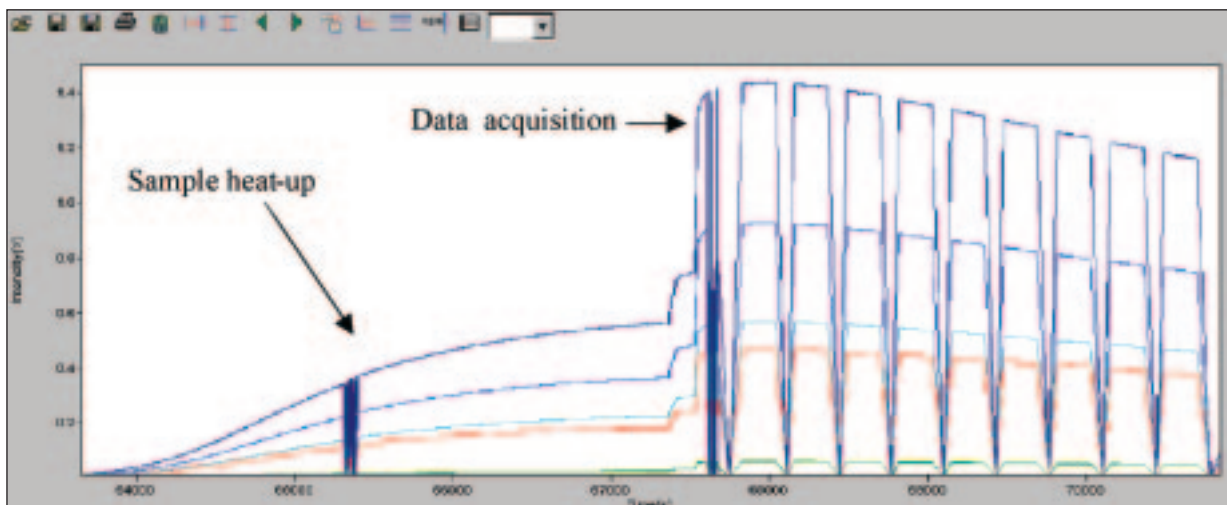
function of the filament material, which can be substantially reduced by covering the filament with, for instance, barium.

For osmium analyses, platinum filaments are required. Because platinum very often contains considerable amounts of osmium, rhenium and tungsten, it is essential to use ultra pure platinum material.

Ion currents can be significantly increased by oxygen introduction during sample heat-up and data collection via a viscous flow oxygen bleed inlet system.

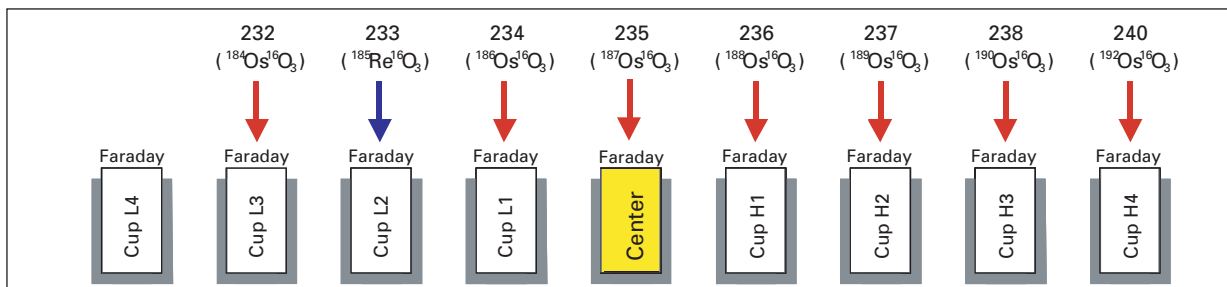
Isotopic fractionation corrections are made using the constant ratio of $^{192}\text{Os}/^{188}\text{Os}$. Because OsO_3^- ions are formed and collected, a correction for the oxygen isotope contribution is required as well.

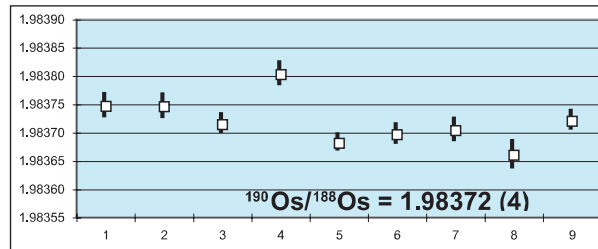
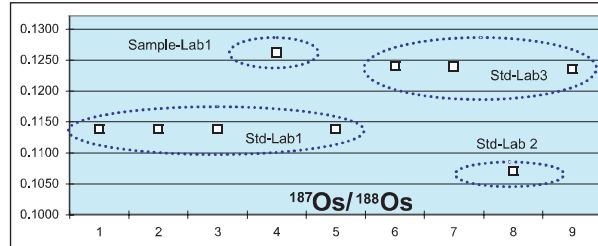
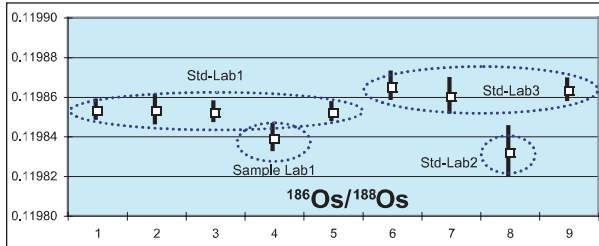
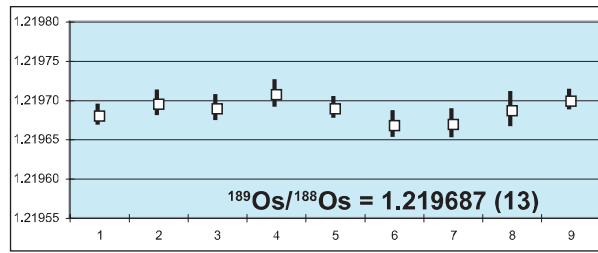
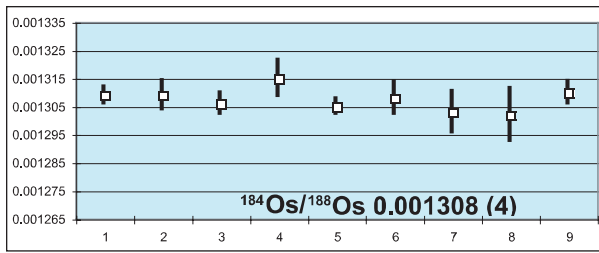
Sample amounts in the low nanogram range can be easily analyzed in multicollector Faraday cup operation mode. Using the advantages of single collector or even multiple collector ion counting, sample amounts in the picogram range can be analyzed with excellent precision.



Isotope Traces of an Osmium Measurement

Schematic of the Multi-Collector Configuration





Analysis Parameter

Samples: 1-35 ng loaded onto a platinum filament together with barium nitrate (Ref: Papanastassiou, Sharma, Brandon et al.)

Sample Warm-up: performed very slowly while introducing oxygen via bleed system, filament currents around 2000 mA, intensities 500-2500 mV

Data Acquisition: Negative ion mode at 10 kV, static Faraday ion collection mode, channel gains calibrated once. Typically collected 300 single data with integration times of 10 s each. Rhenium interference controlled by measuring ¹⁸⁵Re (m/e=233). Isotopic fractionation correction using 240/236=3.092203, followed by oxide correction.

All final data corrected for outliers by 2-sigma-criterion. Total sample analysis time 2-3 hours including sample warm-up.

Summary of Different Osmium Analyses

| RUN # | SAMPLE | ¹⁸⁴ Os/ ¹⁸⁸ Os | 1SE | ¹⁸⁶ Os/ ¹⁸⁸ Os | 1SE | ¹⁸⁷ Os/ ¹⁸⁸ Os | 1SE | ¹⁸⁹ Os/ ¹⁸⁸ Os | 1SE | ¹⁹⁰ Os/ ¹⁸⁸ Os | 1SE |
|--|----------------------------|--------------------------------------|-----|--|-----|--------------------------------------|-----|---|-----|--------------------------------------|-----|
| 1 | STD (Lab#1) 35ng #1 | 0.001309 | 1 | 0.119853 | 2 | 0.113807 | 2 | 1.219680 | 5 | 1.983747 | 10 |
| 2 | STD (Lab#1) (3-5ng) #2 | 0.001309 | 3 | 0.119853 | 3 | 0.113799 | 3 | 1.219695 | 7 | 1.983746 | 10 |
| 3 | STD (Lab#1) (3-5ng) #3 | 0.001306 | 2 | 0.119852 | 2 | 0.113797 | 3 | 1.219689 | 7 | 1.983715 | 7 |
| 4 | Os AP-5 Sample (Lab#1) #4 | 0.001315 | 3 | 0.119839 | 3 | 0.126231 | 3 | 1.219707 | 7 | 1.983803 | 9 |
| 5 | STD (Lab#1) 35ng (2.6V) #5 | 0.001305 | 1 | 0.119852 | 2 | 0.113796 | 2 | 1.219689 | 6 | 1.983682 | 6 |
| 6 | STD (DT-FMAT) #1 | 0.001308 | 3 | 0.119865 | 3 | 0.124075 | 5 | 1.219668 | 7 | 1.983697 | 8 |
| 7 | STD (DT-FMAT) #2 | 0.001303 | 4 | 0.119860 | 4 | 0.123972 | 8 | 1.219669 | 8 | 1.983704 | 9 |
| 8 | STD (Lab#2) #1 | 0.001302 | 5 | 0.119832 | 6 | 0.107010 | 16 | 1.219687 | 10 | 1.983660 | 11 |
| 9 | STD (DT-FMAT) #3 | 0.001310 | 2 | 0.119863 | 2 | 0.123541 | 2 | 1.219699 | 5 | 1.983721 | 7 |
| Averaged SE(int) | | | 2.6 | | 3.1 | | 5.0 | | 7.0 | | 8.6 |
| Mean | | 0.001307 | | 0.119853 | | 0.113800 | | 1.219687 | | 1.983719 | |
| SD (external) | | 0.000004 | | 0.000001 | | 0.000005 | | 0.000013 | | 0.000042 | |
| RSD (external) | | 0.30 % | | 4.8 ppm | | 43.9 ppm | | 10.7 ppm | | 21.2 ppm | |
| Normalization using: ¹⁹² OsO ₃ / ¹⁸⁸ OsO ₃ = 3.092203 | | | | Above mean values and statistics are of samples #1,2,3,5 (Laboratory #1) | | | | Oxide correction using: ¹⁷⁰ O/ ¹⁶ O = 0.0003708, ¹⁸ O/ ¹⁶ O = 0.002045 | | | |

Note: Different samples vary in ¹⁸⁶Os and ¹⁸⁷Os abundances. The gray shaded fields in the table indicate that the statistics have only been calculated based on the samples, which appear shaded in the table as well.

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